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G. W. Gokel, Crown Ethers and Cryptands., Royal Society of Chemistry, Cambridge, U.K., 1991. Key Words: Lariat ethers: from simple sidearms to supramolecular systems ...

### Author: G. W. Gokel - organic-chemistry.org

H. Dupont Durst received his BS from LSU in 1965, engaging in undergraduate research, with Professor Bill Patrick, on the inorganic chemistry of soils. Durst describes "Dr Pat" as an "incredible man and a tremendous mentor.". Durst went on to earn an MS in Biochemistry at LSU, completing a thesis under the direction of Gerald Risinger, "On the Mechanism of the Thiamine Catalyzed Reductive Dimerization of Farnesyl Pyrophosphate to Squalene.".

### Dupont Durst | LSU Chemistry

With George Gokel he co-authored the textbook "Experimental Organic Chemistry." From 1979-84, he was a faculty member at the University of Puerto Rico in Rio Piedras, where research interests expanded to include the isolation and structure determination of marine natural products, including toxins associated with seafood poisoning.

### H. Dupont Durst Graduate Student Award in Organic Chemistry

EXPERIMENTAL ORGANIC CHEMISTRY BY JAMES F. NORRIS Professor of Organic Chemistry, Massachusetts Institute of Technology; Author of "The Principles of Organic Chemistry," "A Textbook of Inorganic Chemistry for Colleges," and Joint Author of "Laboratory Exercises in Inorganic Chemistry" SECOND EDITION SECOND IMPRESSION TOTAL ISSUE, 38,000

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2 3. Literature Adapted from: Gokel, H. D.; Durst, G. W. Experimental Organic Chemistry; McGraw-Hill, New York, 1980; pp 344-348.

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melting point of both sets of crystals. Literature melting point = 136-137°C. (laboratory adapted from the procedures in "Experimental Organic Chemistry" by Durst and Gokel and "Experimental Methods in Organic Chemistry" by Moore, Dalrymple and Rodig.)

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A solution of 30% aqueous hydrogen peroxide (9ml,85.5mm) and formic acid (16ml,88%)is added to a solution of isoeugenol (8.1g,50mm) in formic acid (4ml). The reactionmixture is stirred at 35-40degC under nitrogen atmosphere for 3 hours.

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