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Learning Answer Key - PDF ...**

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Gizmo Warm-up A calorimeter is an insulated container filled with a liquid, usually water. When a hot object is placed in the calorimeter, heat energy is transferred from the object to the water and the water heats up. Calorimeters can be used to find a substance's specific heat capacity.

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Student Exploration- Calorimetry Lab (ANSWER KEY)

Calorimetry Virtual Lab - BetterLesson
Calorimetry Lab Gizmo Answers Activity
C. Calorimetry Lab Gizmo Answers
Activity C. Discover how animals, plants,
and sunlight interact to maintain a
balance of gases in Earth's atmosphere.
0°C is added to 25. 0 °C and the

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copper's Temp is 90 °C. dic This class can parse, analyze words and interprets sentences. On the SIMULATION pane, select Copper.

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You will use the Calorimetry Gizmo laboratory™ which determines the

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specific thermal abilities of various substances. 1. On the SIMULATION pane, select Bronze. Use the slider to set its Mass to 200 g. Set the volume of water to 200 g. Check that the water temperature is set at 30.0 °C and the copper temperature is 90 °C.

Student exploration calorimetry lab

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Investigate how calorimetry can be used to find relative specific heat values when different substances are mixed with water. Modify initial mass and temperature values to see effects on the system. One or any combination of the substances can be mixed with water. A

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dynamic graph (temperature vs. time)
shows temperatures of the individual
substances after mixing.

Calorimetry Lab Gizmo : ExploreLearning

Student Exploration- Calorimetry Lab
(answers) Description Student
Exploration: Calorimetry Lab Vocabulary:

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calorie, calorimeter, joule, specific heat capacity Prior Knowledge Questions (Do these BEFORE using the Gizmo.) The Latin word calor means “heat,” and meter comes from the Greek word meaning “to measure.”.

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Student Exploration: Calorimetry Lab
Vocabulary: calorie, calorimeter, joule, specific heat capacity
Prior Knowledge Questions (Do these BEFORE using the Gizmo.)
1. The Latin word calor means “heat,” and meter comes from the Greek word meaning “to measure.” What do you think a calorimeter does? Measures heat
2. Where have you heard the word

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calorie before? What do you think a calorie is?

Calorimetry Lab SE - Student Exploration Calorimetry Lab ...

Student Exploration- Calorimetry Lab
(ANSWER KEY) Gizmo Warm-up A
calorimeter is an insulated container
filled with a liquid, usually water When a

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hot object is placed in the calorimeter,
heat energy is transferred Page 2/8.
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Correct Answer: C. The final temperature
of the lead-water system will be lower

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than the final temperature of the copper-water system. A blacksmith heats a 1,540 g iron horseshoe to a temperature of 1445°C before dropping it into 4,280 g of water at 23.1°C .

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Calorimetry Lab. Launch Gizmo.
Investigate how calorimetry can be used

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to find relative specific heat values when different substances are mixed with water. Modify initial mass and temperature values to see effects on the system. One or any combination of the substances can be mixed with water.

**Calorimetry Lab Gizmo : Lesson Info
: ExploreLearning**

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This project was created with Explain Everything™ Interactive Whiteboard for iPad.

Calorimetry Gizmo Part 2 Help - YouTube

Gizmo Warm-up. A calorimeter is an insulated container filled with a liquid, usually water. When a hot object is

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placed in the calorimeter, heat energy is transferred from the object to the water and the water heats up. Calorimeters can be used to find a substance's specific heat capacity. You will use the Calorimetry LabGizmo™ to determine the specific heat capacities of various substances.

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Student Exploration: Calorimetry Lab

Student Exploration Calorimetry Lab
Gizmo Answers Calorimetry Lab Gizmo
to determine the specific heat capacities
of various substances On the
SIMULATION pane select Copper Use the
slider to set...

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Calorimetry Lab Calorimeters can be used to find a substance's specific heat capacity. You will use the Calorimetry Lab Gizmo™ to determine the specific heat capacities of various substances. 1.

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On the SIMULATION pane, select Copper.
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to determine the specific heat capacities of various substances. 1. On the SIMULATION pane, select Copper. Use the slider to set its Mass to 200 g. Set the Water mass to 200 g. Check that the Water temp is set to 30.0 °C and the copper's Temp is 90 °C.

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