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Gases Study Guide Properties of a Gas. Gas Balloon. ... A gas is a state of matter. The particles that make up a gas can range from... Pressure. Pressure is a measure of the amount of force per unit area. The pressure of a gas is the amount of force the... Temperature. Temperature is a property of ...

Chemistry Study Guide for Gases - ThoughtCo

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 $R = 0.0821 \text{ liter}\cdot\text{atm}/\text{mol}\cdot\text{K}$ ($P = \text{atm}$, $V = \text{L}$ and $T = \text{K}$)
 $R = 8.3145 \text{ J}/\text{mol}\cdot\text{K}$ (Pressure x Volume is energy, $T = \text{K}$)
 $R = 8.2057 \text{ m}^3 \cdot \text{atm}/\text{mol}\cdot\text{K}$ ($P = \text{atm}$, $V = \text{cubic meters}$ and $T = \text{K}$)
 $R = 62.3637 \text{ L}\cdot\text{Torr}/\text{mol}\cdot\text{K}$ or $\text{L}\cdot\text{mmHg}/\text{mol}\cdot\text{K}$ ($P = \text{torr}$ or mmHg , $V = \text{L}$ and $T = \text{K}$)
The ideal gas law works well for Page 4/26

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Gas consists of large numbers of tiny particles that are far apart relative to their size. 2. Collisions between gas particles and container walls are perfectly elastic collisions. (Elastic Collision-> a collision in which there is no loss of total kinetic energy.)

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In the kinetic molecular model, gas particles are in constant random motion, straight line motion, they are of negligible volume, they have no forces of attraction, and they have elastic collisions What does the temperature determine? Average kinetic energy of particles- must be expressed in Kelvin

Chemistry Gas Laws Study Guide Flashcards | Quizlet

Waves Unit Study Guide KEY 2 Figure 1 Figure 2 Use Figure 1 and Figure 2 above to answer questions 10. Which wave(s) is a transverse wave? Figure 2 11. Which wave(s) is a compressional wave? Figure 1 12. Which wave(s) is an electromagnetic wave? Figure 2 13. Which wave(s) could use a medium to transfer energy? Figure 1 and Figure 2 14.

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Waves Unit Study Guide KEY - Troup County School District

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Gases Unit Study Guide Answers The average kinetic energy of a gas is expressed by the formula: $KE_{ave} = 3RT/2$ where KE_{ave} = average kinetic energy R = ideal gas constant T = absolute temperature The average velocity or root mean square

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Matter Unit Study Guide [1] - Polk County School District Waves

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Unit Study Guide KEY 3 25. Explain why you can see lightning before you hear thunder. Light travels faster than sound 26. Sort the following according to how fast sound will travel through the medium: liquid, solid, gas. Study guide - Answers in Genesis

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Earth's Atmosphere Unit Test Study Guide Know the definitions of the following terms: Climate- the characteristic weather conditions in an area over a long period of time. Air mass- Large body of air. Covers thousands of miles. Fossil fuels-Fuels formed from the remains of prehistoric organisms that are burned for energy.

Earth's Atmosphere Unit Test Study Guide

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usual for many people. However, there are still many people who plus don't later reading. This is a problem. But,

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Gas Laws STUDY GUIDE Due: February 12th Units of Measurement: For the following questions, use the following answer choices to indicate what each unit of measurement is used to measure. A. Pressure B. Volume C 1. K 4. kPa A 2. atm 5. L 3. mL 6. °C C. Temperature 'A 7. A 8.

Gas Laws STUDY GUIDE Due: February 12th

Chemistry Gases Unit Study Guide The ideal gas law is expressed by the formula $PV = nRT$ where P = pressure V = volume n = number of moles of gas R = ideal gas constant T = absolute temperature The value of R depends on the units of pressure, volume and temperature. Chemistry Study Guide for Gases - ThoughtCo

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Thermosphere Increases with altitude Contains Ions -charged gases
Mesosphere Decreases with altitude Burns up meteors
Stratosphere Increases with altitude Contains the Ozone Layer
Troposphere Decreases with altitude The layer you live in and where weather occurs
5. How are the layers of the atmosphere broken up/determine?

Atmosphere Study Guide Quiz 1 - Northeastern York School ...

Unit Study Guide: Study Guide: States of Matter Subpages (5):
Lesson 1: States of Matter Lesson 2: State Changes Lesson 3:
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Unit 2: States of Matter - SMHS Physical Science

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Gases- Gases have no shape or volume of their own. □
Examples: oxygen, helium, steam Atoms in a solid are very close together and hardly move. Atoms in a liquid are more spread out and move a little. Atoms in a gas are very far apart and move very quickly.

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