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22CHAPTER Study  
Guide Key Concepts  
22.1 Hydrocarbons •  
Because carbon has  
four valence electrons,  
car-bon atoms always  
form four covalent  
bonds. • The carbon  
atoms in an alkane can  
be arranged in a  
straight chain or in a

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chain that has

branches. • Molecules  
of hydrocarbons, such  
as alkanes, are  
nonpolar molecules.

## 22.2 Unsaturated Hydrocarbons

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A controlled process by  
which hydrocarbons  
are broken down or  
rearranged into  
smaller, more useful  
molecules natural gas

This is typically made

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Answer  
up of 80% methane,  
10% ethane, 4%  
propane, and 2%  
butane

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AnswerHydrocarbons

Study Guide Answers

Hydrocarbons are organic molecules that only contain two elements: carbon and hydrogen. Common hydrocarbons include methane, propane, and



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Answer  
butane. Hydrocarbons  
are the primary  
constituents of fossil...  
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## **22hydrocarbon**

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Hydrocarbon Rings •

Some hydrocarbon  
compounds have a  
carbon chain that is in  
the form of a ring • In a  
benzene molecule, the  
bonding Section

22hydrocarbon

Compound Answer

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Chapter 22

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Guide A controlled process by which hydrocarbons are broken down or rearranged into smaller, more useful molecules natural gas This is typically made up of 80% methane, 10% ethane, 4% propane, and 2% butane Study Guide: Chapter 22 Hydrocarbon

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Answer  
molecules of hydrocarbons, such as alkanes, are nonpolar molecules. the simplest organic compounds contain only carbon and hydrogen. a hydrocarbon in which there are only single covalent bonds.

contains any number of carbon atoms, one after the other, in a chain.

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Chapter 22 Name: 1. A

term for when inflation  
is negative is (A)

deflation (B)

disinflation (C)

deadweight loss (D)

shortage (E)

underemployment 2. A

term for when inflation  
is decreasing (for

example if the rate of  
inflation falls from 5%

to 2% per year) is

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Compounds Make the

following Foldable to

help you organize

information about

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