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Advanced Chemistry with Vernier 19 - 1 Lab: Buffer Preparation and Testing A buffer is a mixture of a weak acid and its conjugate base, or a weak base and its conjugate acid A buffer's function is to absorb acids (H⁺ or H₃O⁺ ions) or bases (OH⁻ions) so that the pH of the system changes very, very little in many systems, buffers are critical

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Experiments | Vernier

Advanced Chemistry with Vernier 30 - 5 Lab #10 PRESTUDY Gases: Charles' Law Name ____ Section ____ 1)(2 points) Convert the following temperatures and express the answers to the proper number of significant figures: a) 2530 °C to K b)23 K to °C 2)

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AP Chemistry | Vernier

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