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Section 2-4 Chemical Reactions and Enzymes(pages 49-53) This section describes what happens to chemical bonds during chemical reactions. It also explains how energy changes affect chemical reactions and describes the importance of enzymes. Chemical Reactions(page 49)

Section 2-4 Chemical Reactions and Enzymes

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Section 2.4 STUDY GUIDE CONTINUED
MAIN IDEA: Chemical reactions release or absorb energy. 6. The of the reactants and products determines whether energy will be released or absorbed during a chemical reaction. 7. Before a chemical reaction can start, must be absorbed by the reactants. The amount that must be absorbed to start the reaction is called the.. 8.

SECTION CHEMICAL REACTIONS 2.4 Study Guide

Section 2.4 Chemical Reactions 53 b a
2.4 Chemical Reactions Iron is an element with many desirable properties. It is abundant, easy to shape when heated, and relatively strong, especially when mixed with carbon in steel. Iron has one main disadvantage. Over time, objects made of iron will rust if they are left exposed to air. The brittle layer of rust

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SECTION 2.4 CHEMICAL REACTIONS

Power Notes Chemical Reaction $6O_2 +$

$C_6H_{12}O_6 \rightarrow 6CO_2 + 6H_2O$ 1. 2. 3.

Bond energy: Chemical equilibrium:

Exothermic: 5. activation energy 6.

Reaction Endothermic: progress 4.

reactants 9. activation energy 7.

products 10. Reaction progress 8.

reactants products 11. reactants;

substances changed by a chemical

reaction

CHEMICAL REACTIONS 2.4 Power Notes

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2.4 Chemical Reactions and Enzymes.

2.4 Chemical Reactions and Enzymes. *

Chemical Reactions A process that changes or transforms one set of

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chemicals into another Mass and energy are conserved Reactants Products 2 kinds: energy releasing (exothermic) and energy absorbing (endothermic) Exothermic Reactions Reaction in which heat is given off (Ex. combustion of fuels) * Endothermic Reactions Reaction in which heat is absorbed (Ex. water is evaporated) * Activation Energy Energy needed to get ...

2.4 Chemical Reactions and Enzymes

Chemical Reactions Everything that happens in an organism is based on chemical reactions. A chemical reaction is a process that changes one set of chemicals into another set of chemicals. The elements or compounds that enter into the reaction are the reactants. The elements or compounds produced by the reaction are the products. Chemical reactions involve changes in the chemical bonds that join atoms in compounds.

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2.4 Chemical Reactions and Enzymes

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chapter 2 section 4 chemical reactions chemistry ...

Chemical reaction. a process that changes, or transforms, one set of chemicals or.... Reactant. elements or compounds that enter into a chemical reactions (th.... 11 Terms. filipekscience. Section 2-4 Chemical Reactions and Enzymes. Chemical Reaction. Activation Energy.

section 2 vocab chapter 4 chemical reactions Flashcards ...

SeCtion 4 Instant Replay The1. bondsof reactants break and then form to produce the products. An2. exothermic reaction gives off more energy than it

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uses, and an endothermic reaction uses more energy than it gives off.

Teacher Notes and Answers

2.4 Chemical Reactions Bonds break and form during chemical reactions. • Chemical reactions change substances into different ones by breaking and forming chemical bonds. – Reactants are changed during a chemical reaction.

2.4 Chemical Reactions

Section 2-4 Chemical Reactions and Enzymes Date _____ (pages 49-53) Key Concepts • What happens to chemical bonds during chemical reactions? • How do energy changes affect whether a chemical reaction will occur? • Why are enzymes important to living things? Chemical Reactions (page 49) 1. What is a chemical reaction?

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Coco Wen 11/15/17 Per.6 Biology 2-4 Section Assessment 1. During chemical

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reactions, the chemical bonds between atoms are changed. 2. Some chemical reactions release energy, and other chemical reactions absorb energy.

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Section 2.4 1. reactants; substances changed by a chemical reaction 2. direction of the reaction 3. products; substances formed during a chemical reaction 4. reactants 5. activation energy 6. total energy released 7. products 8. products 9. activation energy 10. total energy released 11. products Bond energy the amount of energy needed to break ...

Power Notes Answer Key - Chapter 2

Name Period Date Section 2.4 STUDY GUIDE CONTINUED MAIN IDEA: Chemical reactions release or absorb energy. 6. The of the reactants and products determines whether energy will be released or absorbed during a chemical

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reaction.

SECTION 2.4 Study Guide -

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Section R eview 2.4 PartA Completion 1. chemical 2. physical 3. chemical 4. reactants 5. chemical 6. composition 7. conservation of mass 8. mass PartB True-Fals e 9. ST 11. ST 13. NT 10. AT 12. NT PartC Matching 14. e 16. d 18. a 15. b 17. c PartD Qu estions and Problems 19. The products of the reaction are gases, which

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